

Monitoring Chlorine by X-ray Absorption Spectroscopy of PbCl₂-Derived Perovskite Films

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The inclusion of chlorine has been shown to improve carrier diffusion lengths in methylammonium lead triiodide perovskite films (CH₃NH₃PbI_{3-x}Cl_x) [1]. Preparation from PbCl₂ has also demonstrated improved surface coverage and film uniformity as compared to films derived from PbI₂ [2], and it has been suggested that residual chlorine may have an effect on perovskite device electronics [3].

We have investigated the amount of chlorine and its chemical state in PbCl₂-derived perovskite films throughout the annealing process. This was done through the use of in-situ X-ray absorption near edge spectroscopy (XANES) at the chlorine K-edge. A 1:3 mix of PbCl₂:MAI in DMF was spin-coated on FTO/c-TiO₂ in a nitrogen glovebox; in-situ XANES measurements were then performed while the films were annealed in a helium environment. The films were never exposed to ambient air.

We have previously reported that chlorine leaves the film during annealing

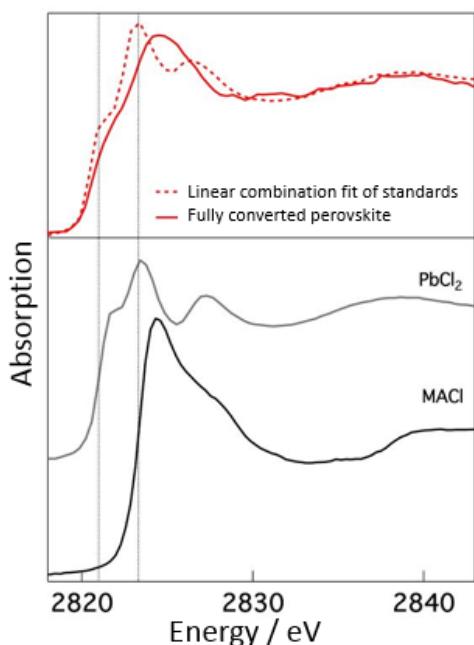


Figure 1: Linear combination fit of the standards MAI and PbCl₂ cannot express perovskite XANES

through sublimation of MAI [4]. Through in-situ XANES, we observe the chlorine content decline consistently throughout the anneal before plateauing above zero. The amount of residual chlorine is estimated by comparison of chlorine and lead absorption edges. CH₃NH₃PbI_{3-x}Cl_x films annealed at a range of temperatures are seen to plateau with a residual amount of chlorine $\sim x=0.05$.

Linear combination fitting of XANES spectra can be used to determine whether an element in a sample matches a set of standards. It has been suggested that residual chlorine may be unconverted PbCl₂ or unevaporated MAI. We have compared the XANES of fully converted perovskite to the standards PbCl₂ and MAI and have found that a linear combination of the standards is unable to express the perovskite XANES. Thus, the residual chlorine in perovskite is not exclusively PbCl₂ and MAI (Figure 1).

References

- [1] S. Stranks *et al.*, *Science* (2013), 342, 341-344.
- [2] Y. Tidar *et al.*, *J. Am. Chem. Soc.* (2014), 136, 13249–13256.
- [3] S. Colella *et al.*, *J. Phys. Chem. Lett.* (2014), 5, 3532-3538.
- [4] E. Unger *et al.*, *Chem. Mater.* (2014), 26, 7158-7165.